

**5<sup>TH</sup> SBSE INTERNATIONAL MEETING**

23 & 24 SEPTEMBRE 2019 - NOVOTEL PARIS-SUD

**SBSE** 

Technical Meeting

# Two decades of Stir Bar Sorptive Extraction

Frank David and Pat Sandra



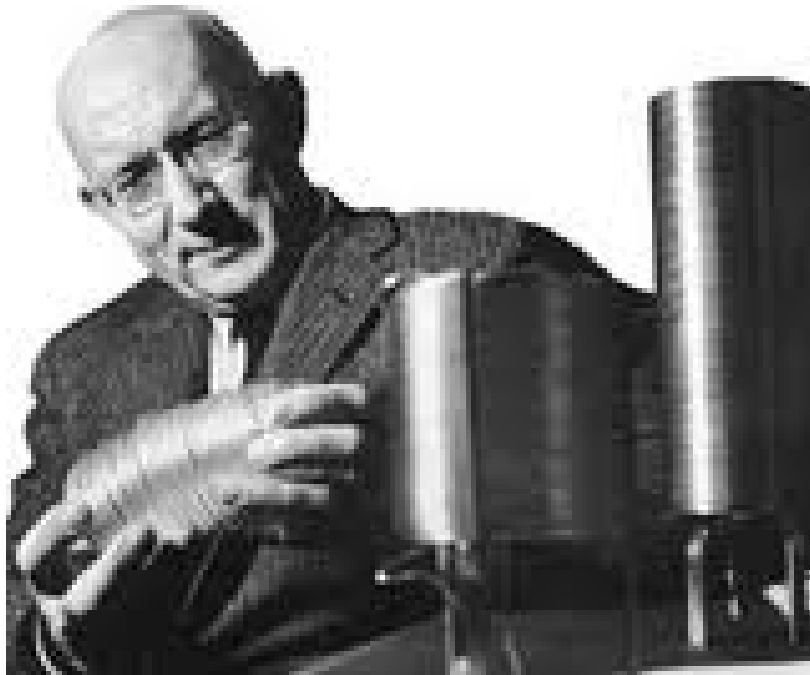
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30 YEARS

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# Capillary GC Milestones



M. Golay, 1959

Static coating:

J. Bouche, M. Verzele (U Gent),  
J. Gas Chromatogr, 6 (1968) 501

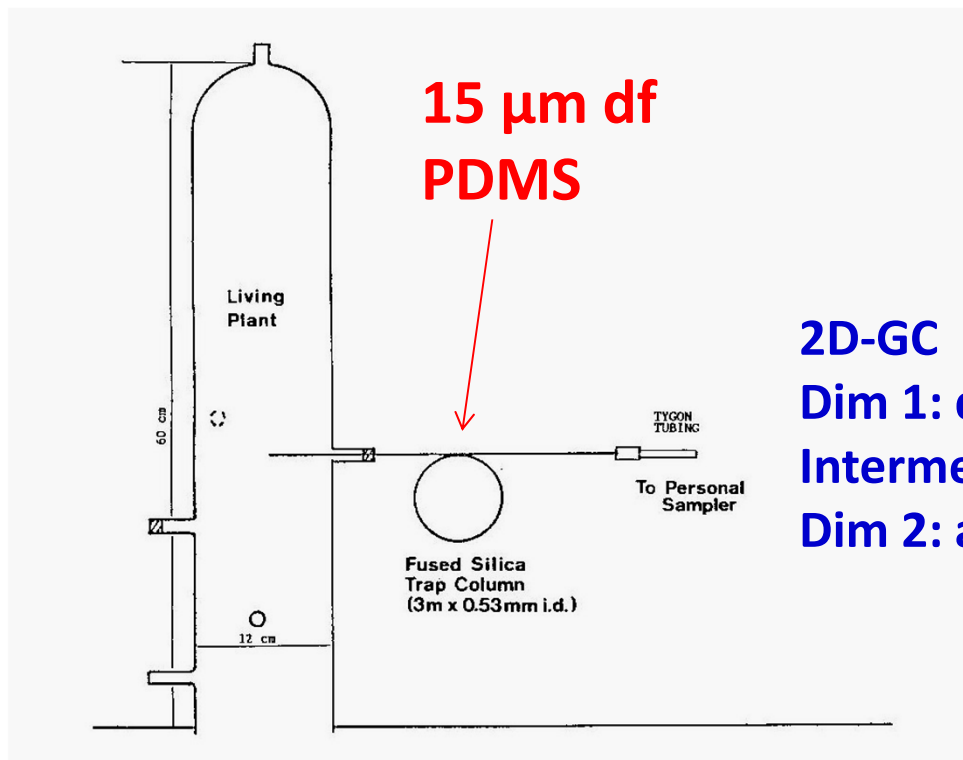
Fused Silica: 1979  
(R. Dandenau, HP)





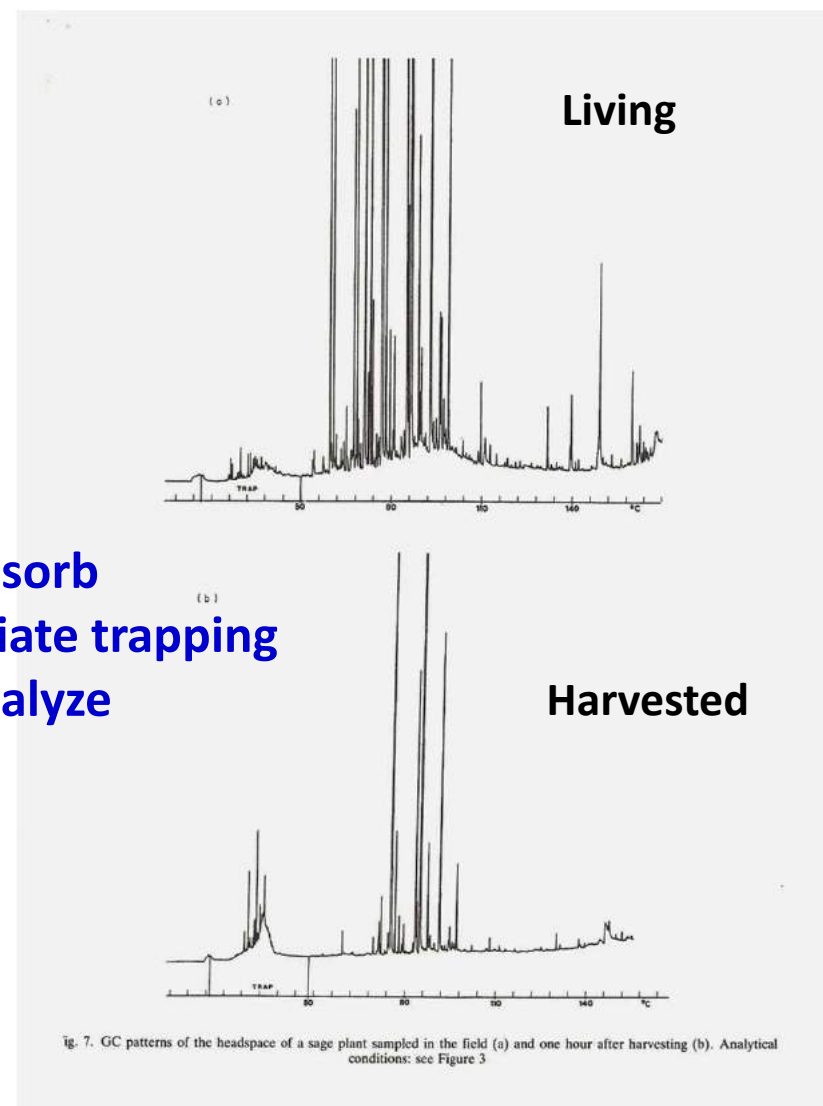
**10° SIMPOSIO INTERNAZIONALE DI-  
(CROMATOLOGIA CAPILLARE) RIVA DEL GARDA 22/25/5/89**

# Open Tubular Trapping of Volatiles



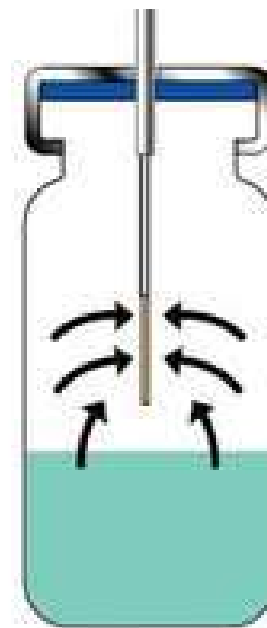
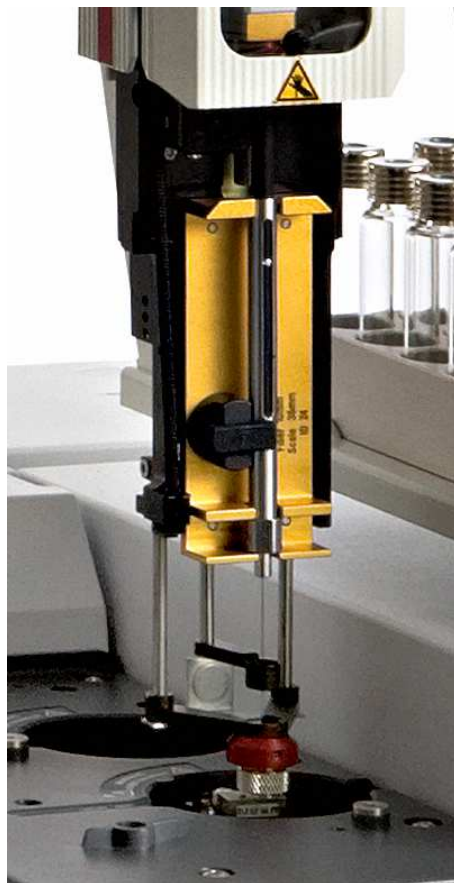
2D-GC  
Dim 1: desorb  
Intermediate trapping  
Dim 2: analyze

Bicchi, D'Amato, David, Sandra  
FFJ 1988



# Solid Phase Micro-Extraction (SPME)

J. Pawliszyn, 1990



$$n = \frac{K_{fs} V_f C_0 V_s}{K_{fs} V_f + V_s}$$

$$(K_{fs} = K_{fa} * K_{as})$$



# Stir Bar Sorptive Extraction

## Sandra, Baltussen and David [1999]

- *Origin: Publication on the SPME extraction of PCBs.*
  - *Authors found very low recoveries for compounds with  $K_{o/w}$  values of up to  $10^{10}$*
  - *Repeating the SPME experiments :*
    - *Similar SPME recoveries were obtained*
    - *However, more than 80 % of the spiked analytes were adsorbed on the (Teflon) stir bar*
- *Idea: Extraction of aqueous samples with a **PDMS coated stir bar***



# Why SBSE?

- **Extraction**: remove from matrix
- **Enrichment**: concentrate
- **Purification**: selective extraction/isolation

# Statement 1

**SBSE**

**=**

**PDMS**



# SBSE phases

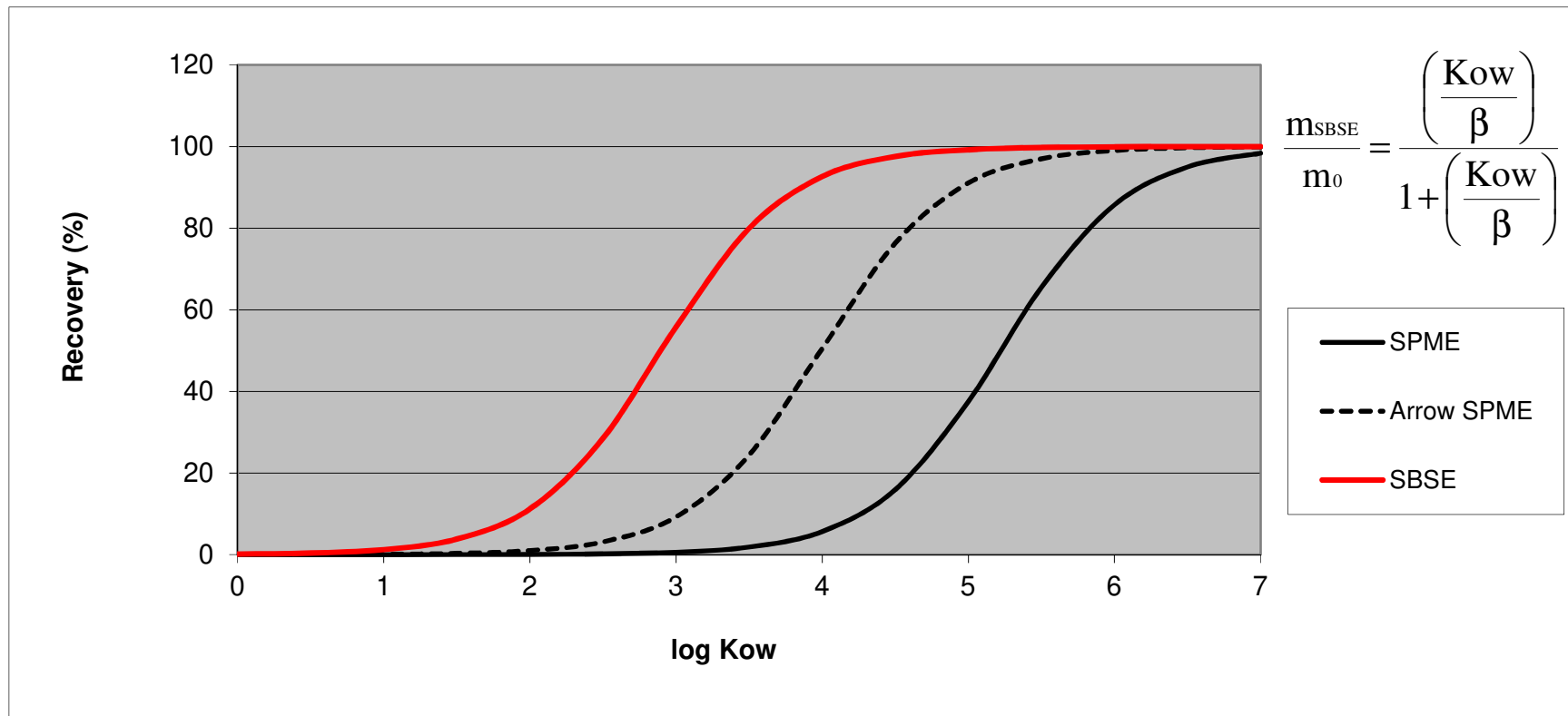
- Best known GC stationary phase (apolar)
- Decomposition products very specific and not related to solutes of interest
- PDMS/water distribution  $\approx$  octanol/water distribution,  $K_{o/w}$  values can be applied
- **WARNING: not all “silicone” material can be used for SBSE**
  - high quality required, low bleed, cross-linker type
- What about other phases for SBSE?
  - Thermal desorption or liquid desorption ?
  - Immersion or Headspace sampling ?
  - More material = bleeding more critical
  - Attempts: polyacrylate, polyurethane, carbon, sol-gel, monoliths, MIPs, RAM,...
  - But: bleeding, liquid desorption only, no significant improvement

## Statement 2

**More extraction phase**

**=**

**More compounds extracted  
(at equilibrium)**



Solute	Log Kow	Recovery (%)		
		SPME	Arrow SPME	SBSE
dichlorvos	1,47	0,02	0,30	3,59
atrazine	2,61	0,24	3,99	33,92
naphthalene	3,30	1,18	16,91	71,54
2,4,6-trichloroanisole	4,11	7,17	56,79	94,20
benzo(a)pyrene	5,99	85,43	99,01	99,92

## Statement 3

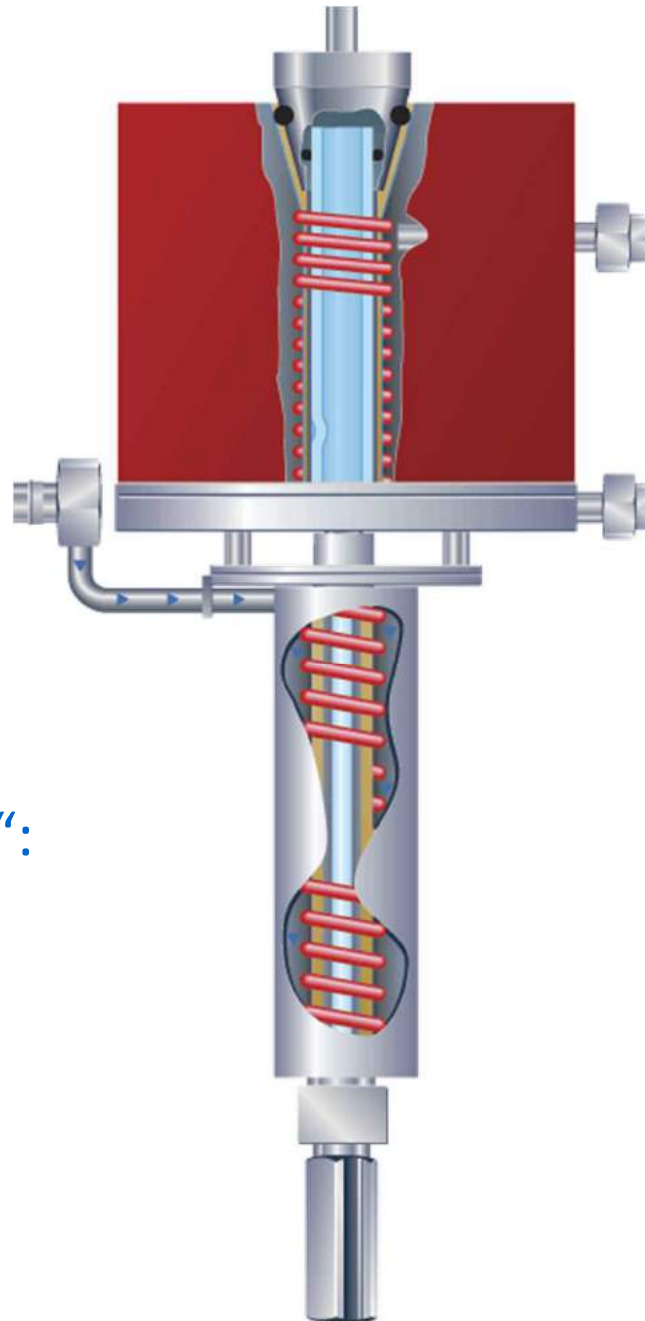
**More extraction phase**

**=**

**More difficult desorption &  
higher bleed**

Thermal Desorber:  
**TDU**

„Cryotrap“:  
**PTV**



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# Statement 4

**SBSE**

**=**

**Green Chemistry**

## Comments on classical sample preparation methods for water analysis

- In most official methods (ISO 6468, ISO 10695, EPA... with LLE or SPE):
  - (very) large sample amounts (250 mL...> 1L)
  - Whole sample bottle extracted (& rinsed) ≠ automation
  - **SPM < 50 mg/L**
  - Some LOQs > EQS levels (endosulfan, heptachlor)
- EU request: “generic” methods for up to 500mg/L SPM

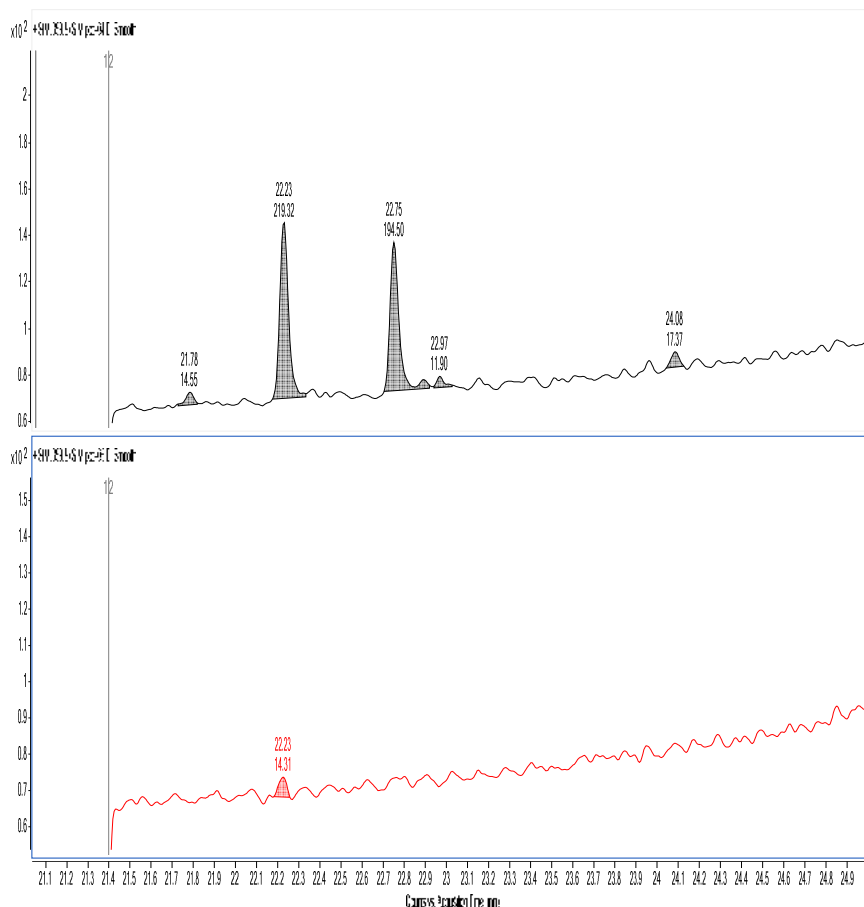
# SBSE in Water Analysis

- Well documented in literature
- Extremely sensitive (pg/L)
- Low external contamination (phthalates)
- **Concern: SPM > low adaptation in regulatory & routine labs**
- Recent research (Gerstel)
  - Dual SBSE extraction – single shot GC-MS/MS method
  - Validated for 100 priority pollutants according to WFD
  - Matrices spiked with 50-100-150 mg/L WEPAL SETOC 745 to simulate SPM



# but... Sampling method should also be adapted!!!

Sampled in 20 mL vial → autosampler

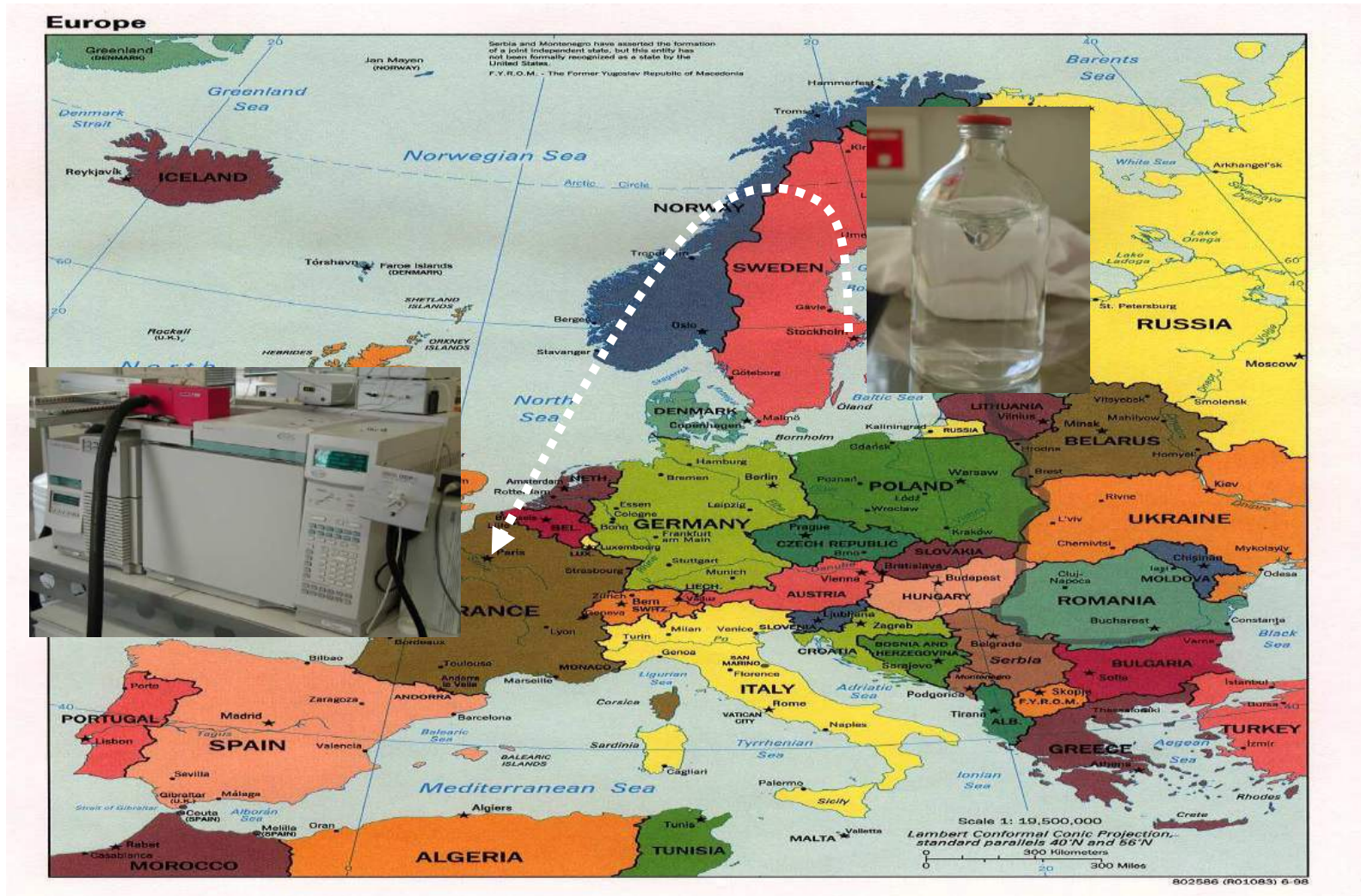


Sampled in 1 L flask → aliquot to 20 mL vial → autosampler



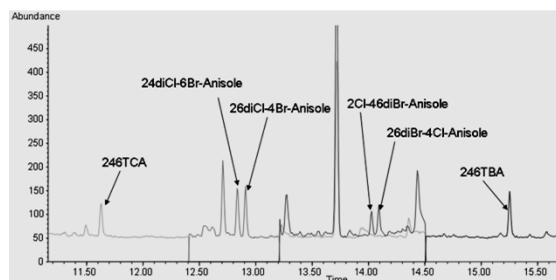
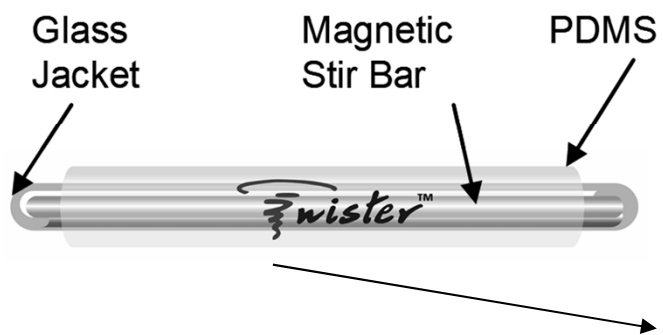
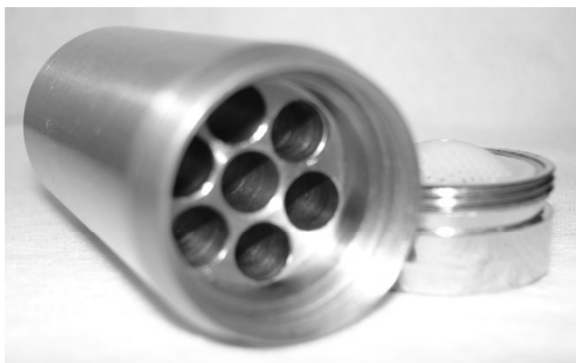
# SBSE-TD-GC-MS - On-site SBSE

D. Benanou, presented at 27<sup>th</sup> ISCC, Riva del Garda, Italy, May 2004

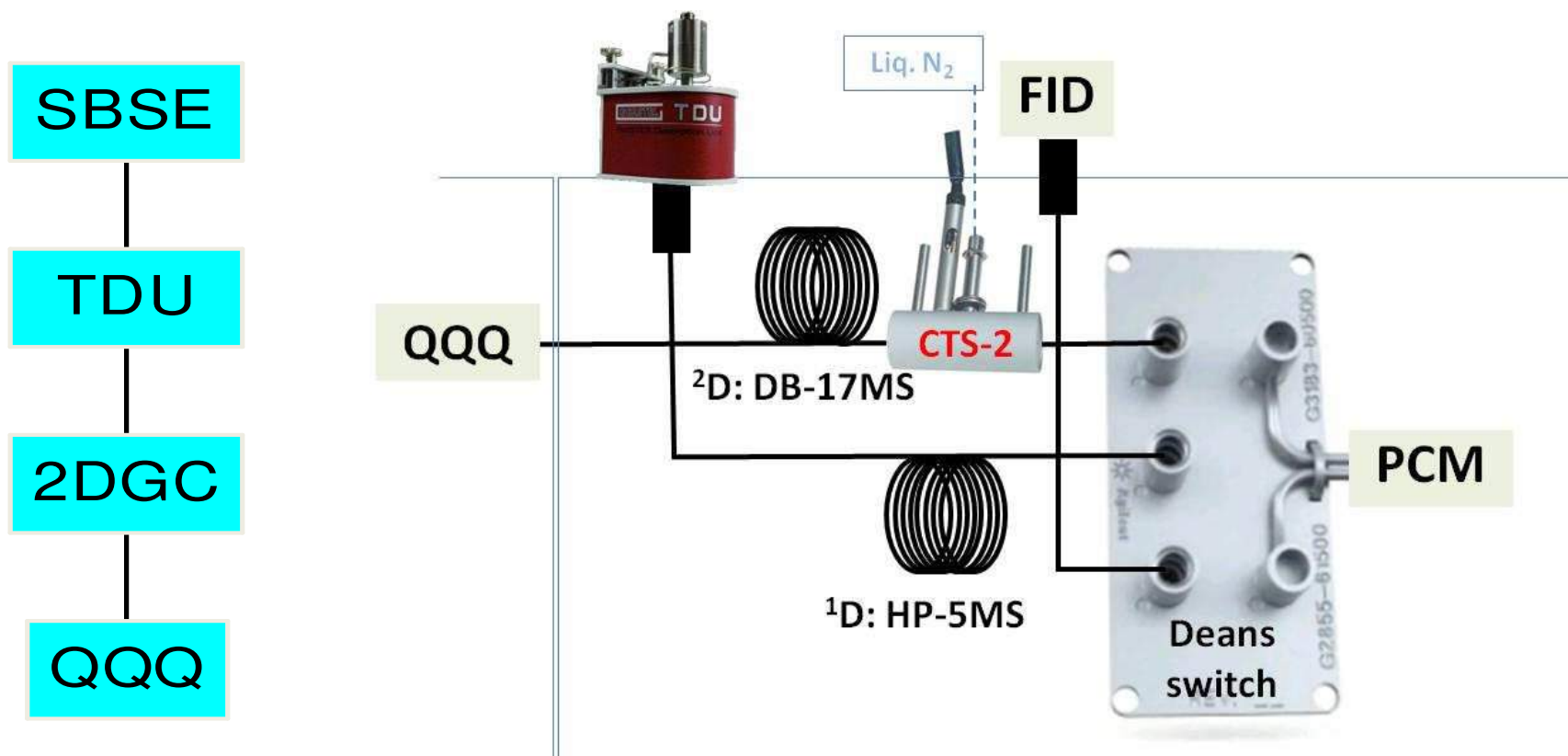


# “On-TAP” sampling

Veolia Environnement, Paris, France



# Determination of Tributyltin in Water Samples at the Quantification Level Set by the European Union (0.2 ng/L)



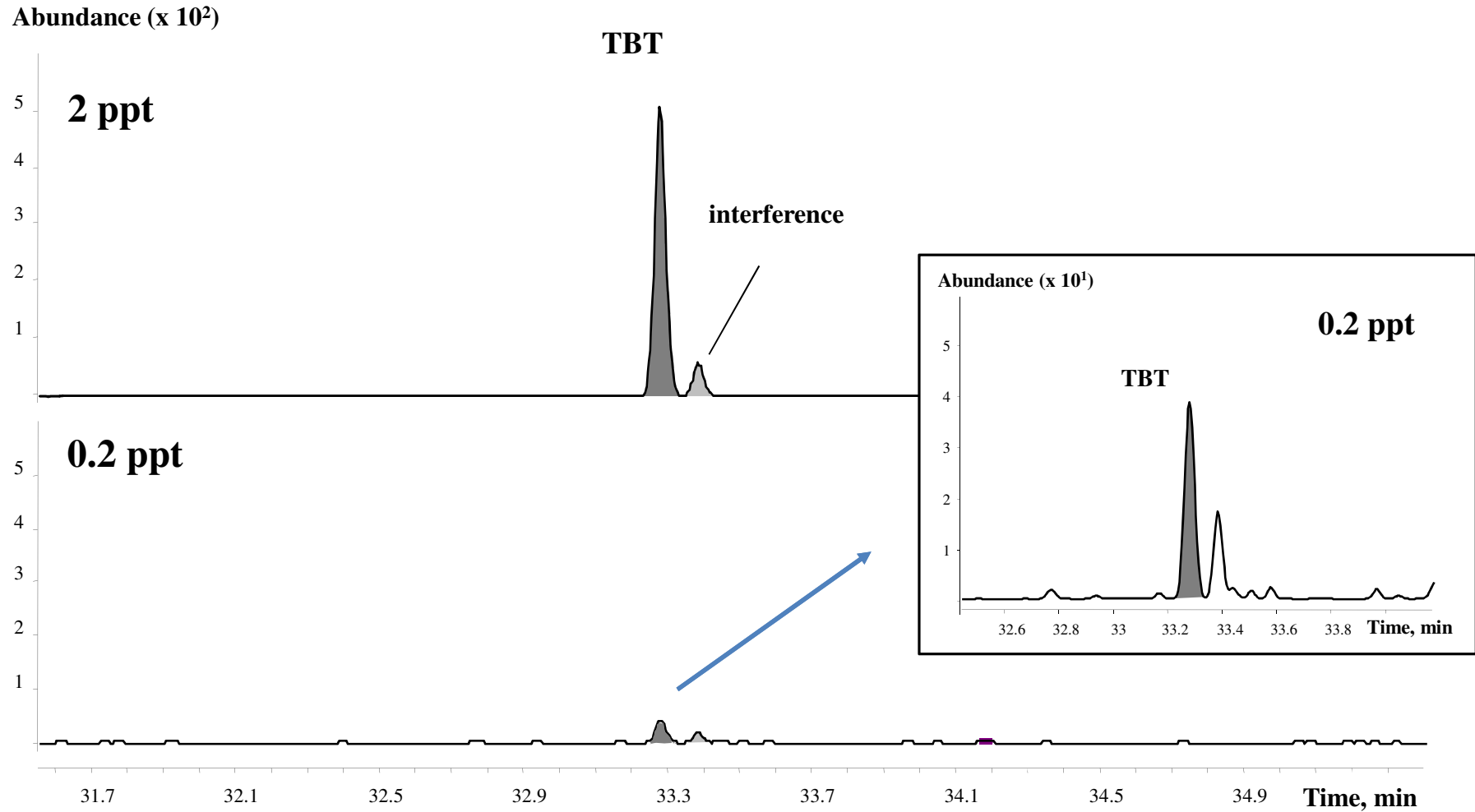
C. Devos, F. David, P. Sandra, J. Chromatography A, 1261 (2012) 151–157

## Sampling procedure – SBSE parameters



- Samples (50 mL) are spiked with deuterated Osn standards
  - 5 mL of buffer (NaOAc/HOAc buffer pH = 5.3) is added
  - 500  $\mu\text{l}$  of 1 %  $\text{NaBEt}_4$  is added for derivatization
  - 2 cm x 0.5 mm  $d_f$  Twisters
  - The Twister is stirred for 2 hours into 50 mL of water sample
- Simple sample preparation – limited glassware and lab material needed**

# GC-GC-MS/MS: TBT resolved from interferences



# Towards modern methods for water analysis

- GC-MS/MS & LC-MS/MS are recommended methods
  - Remove “old” methods: GC-NPD, GC-ECD, LC-UV, LC-FLD
- Encourage (or at least allow) new sample preparation methods
  - → Green chemistry
- Define max SPM level: 50 mg/L > 500 mg/L???
- 95% French water samples < 50 mg/L

# Statement 5

**SBSE**

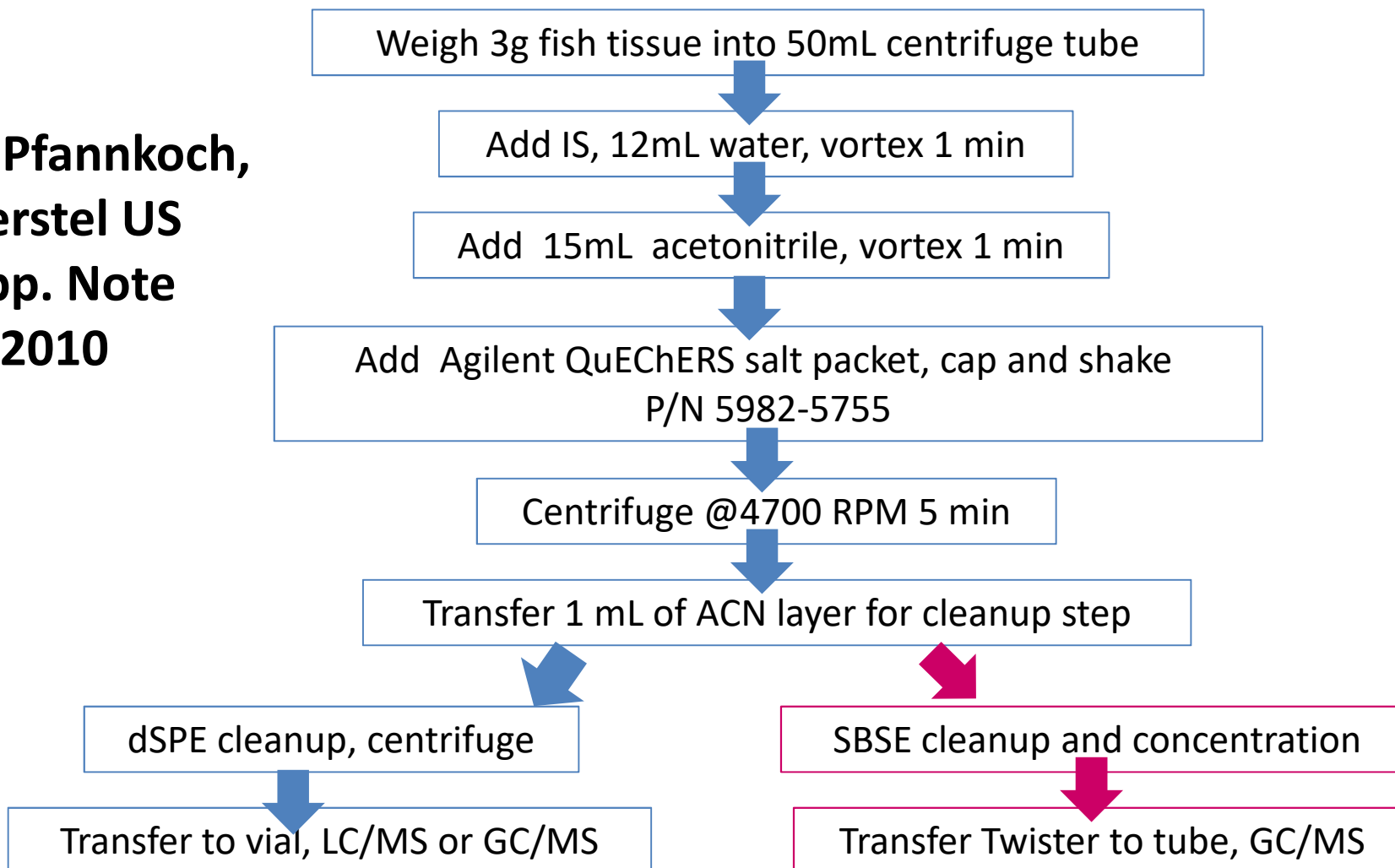
**≠**

**Only extraction of water  
samples**

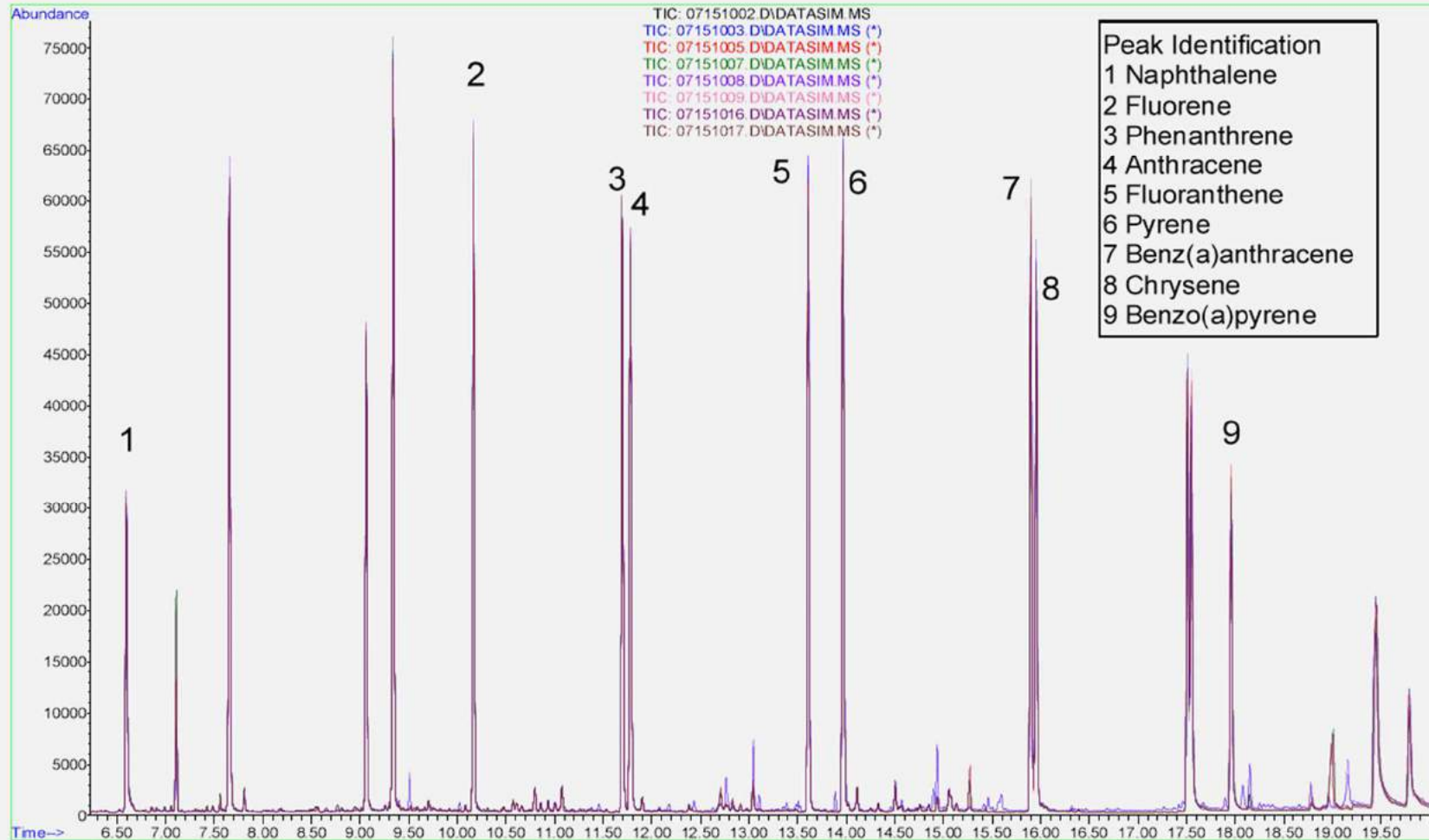


# Determination of PAHs in Sea Food

**E. Pfannkoch,  
Gerstel US  
App. Note  
6/2010**

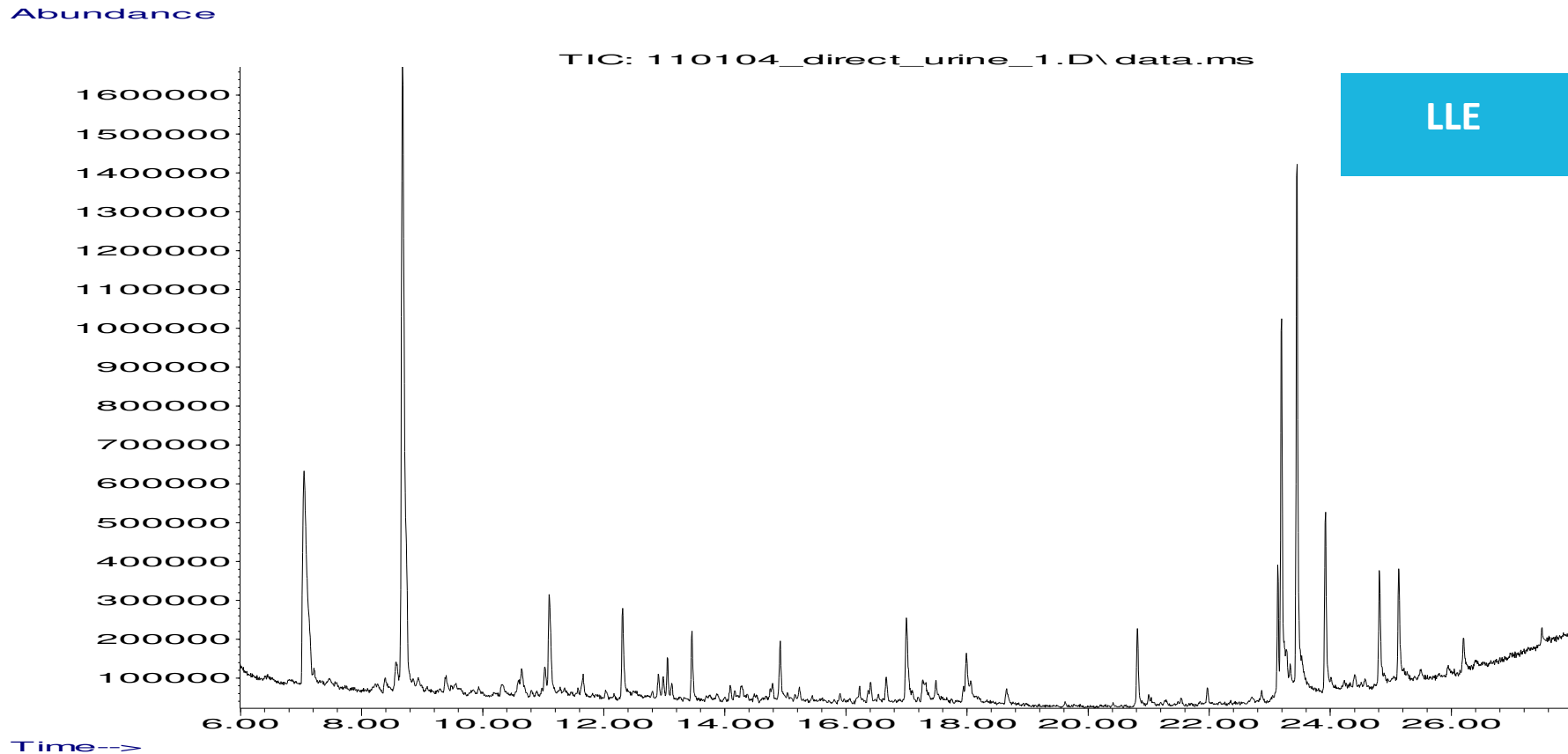


PAH Retention time stability  
25ppb in tissue (croakers, oysters, red snapper)  
18 runs overnight  
8 chromatograms overlaid



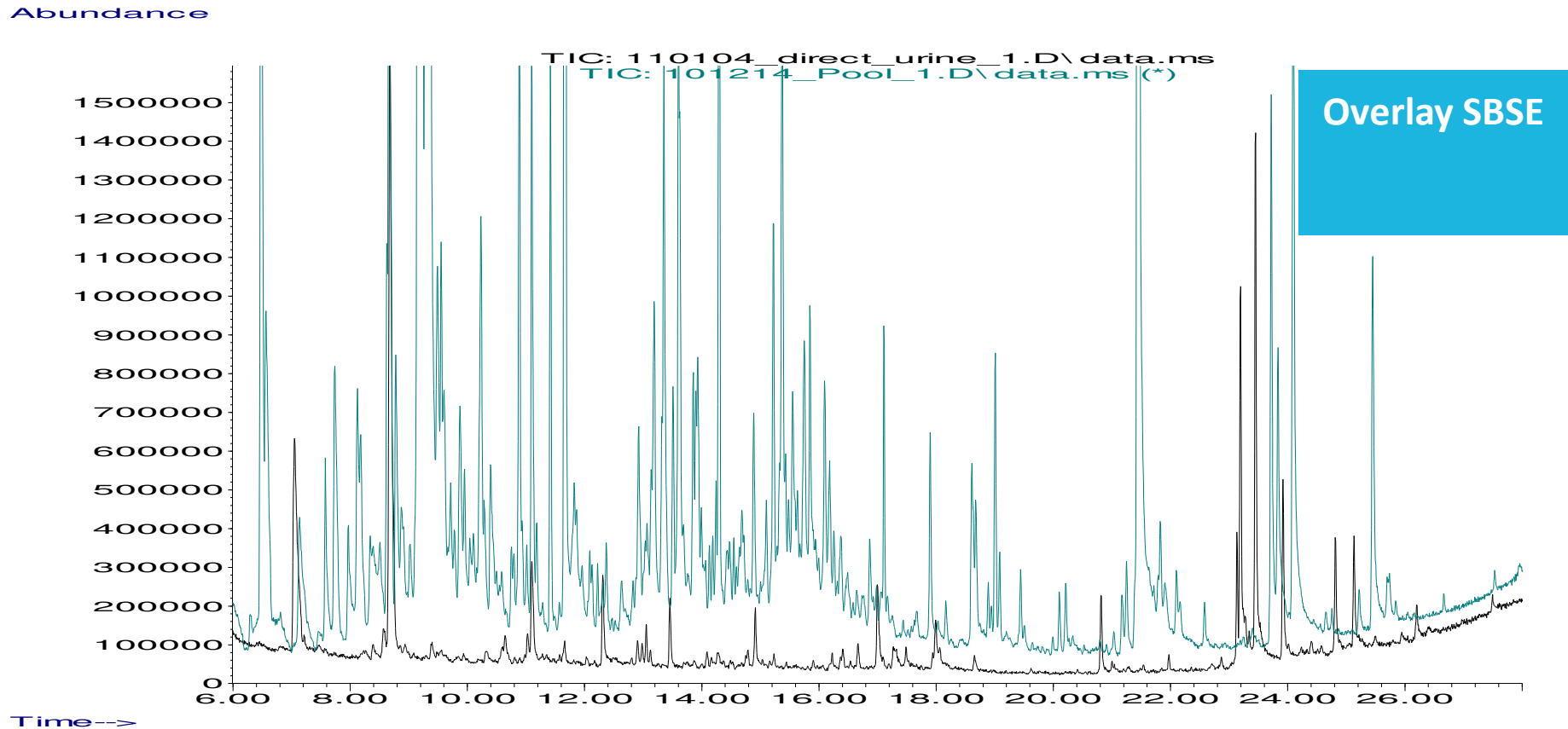
# Body fluid analysis (GC-QTOF)

- Analysis of 1 mL urine



# Comparison SBSE and LLE

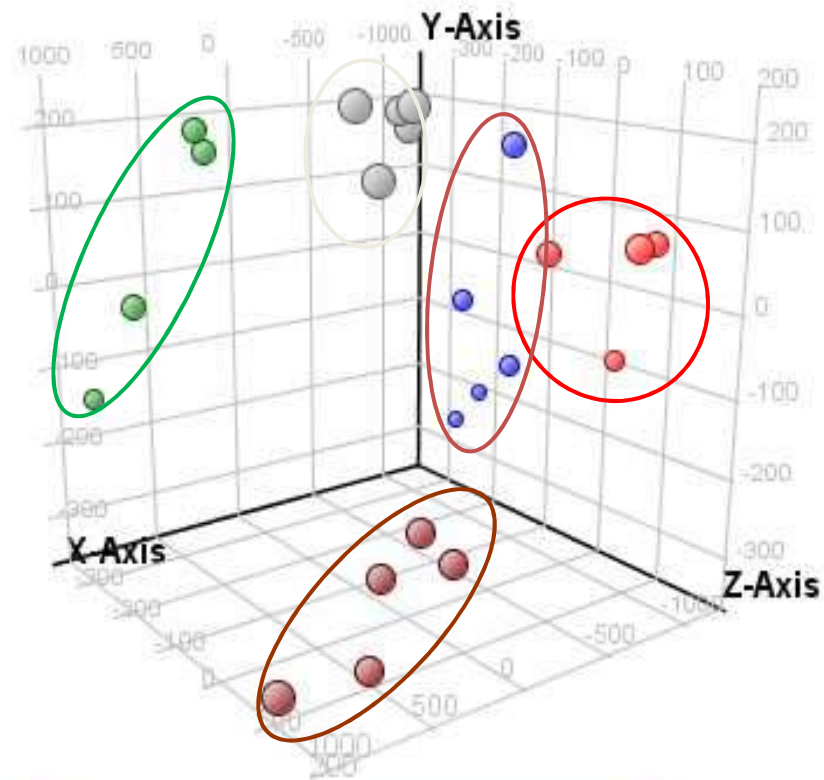
- Analysis of 1 mL urine



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# Urine analysis: 4 individuals x 5 samples + QC



Controle

X-Axis: Component 1 (60.72%)

Y-Axis: Component 2 (16.8%)

Z-Axis: Component 3 (13.2%)



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# The Future of Stir bar Sorptive Extraction and Related Techniques



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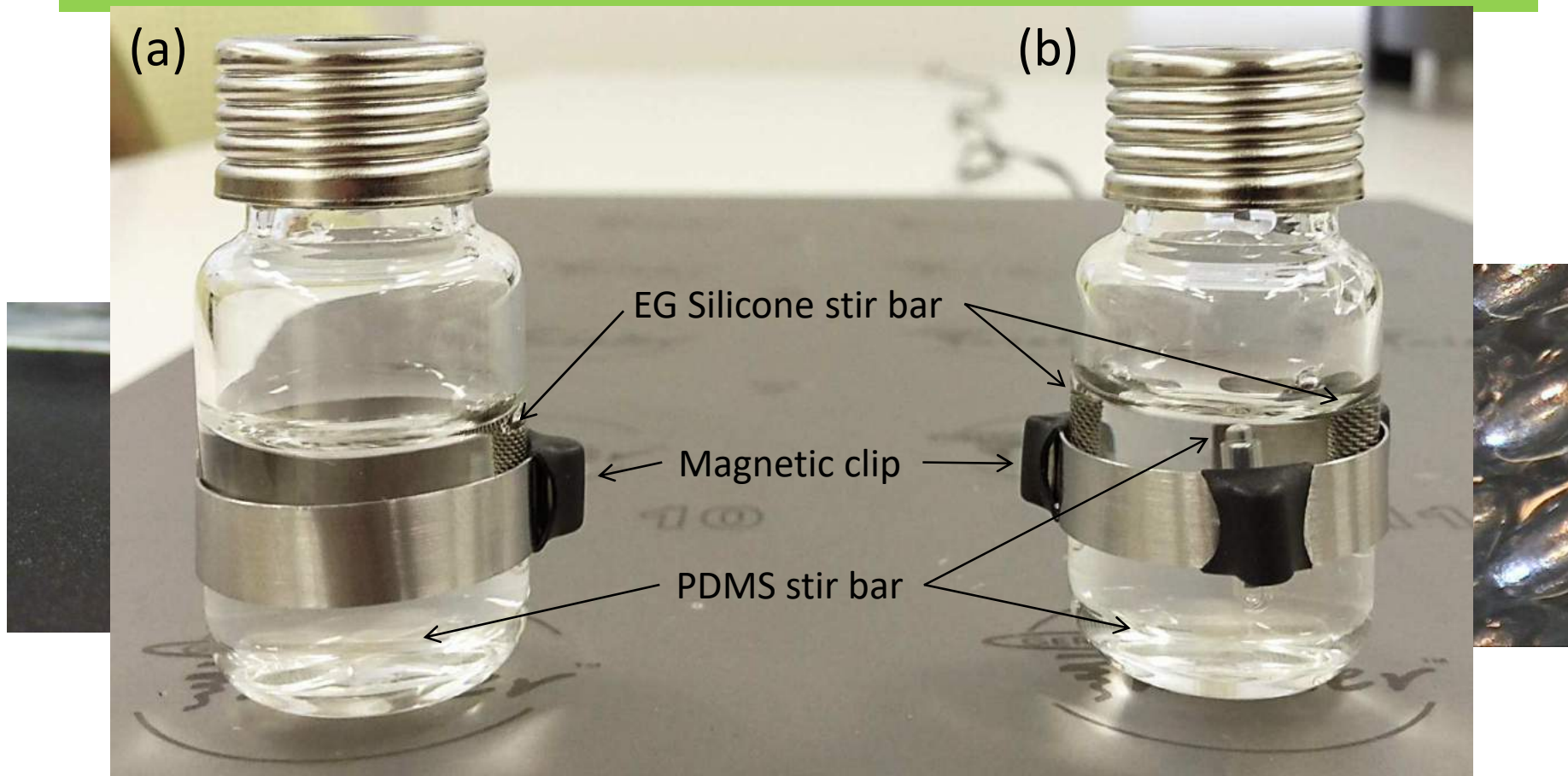
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# Development of Ethylene glycol modified silicone “EG Silicon Twister”

(a)

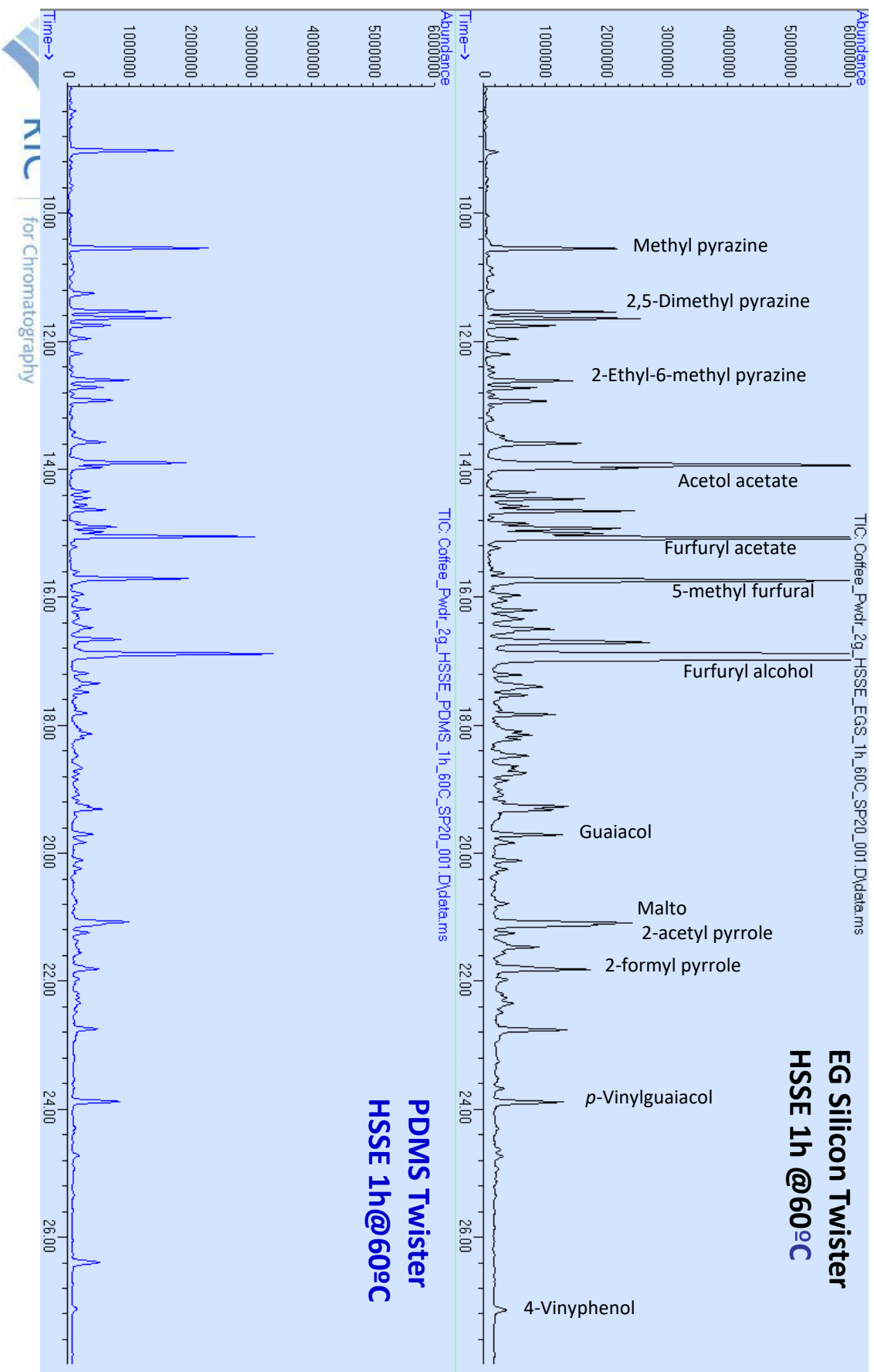
(b)



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# Comparison between EG Silicon Twister and PDMS Twister for HSSE of coffee powder





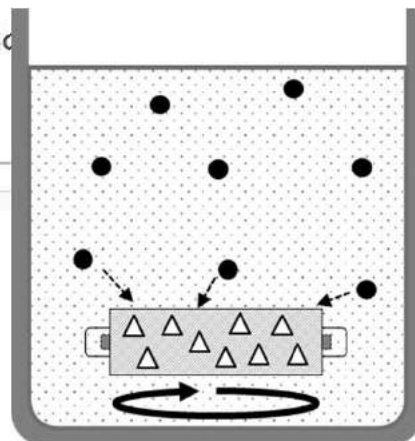
# Solvent-assisted stir bar sorptive extraction by using swollen polydimethylsiloxane for enhanced recovery of polar solutes in aqueous samples: Application to aroma compounds in beer and pesticides in wine

Nobuo Ochiai<sup>a</sup>, Kikuo Sasamoto<sup>a</sup>, Frank David<sup>b</sup>, Pat Sandra<sup>b</sup>  
+ Show more

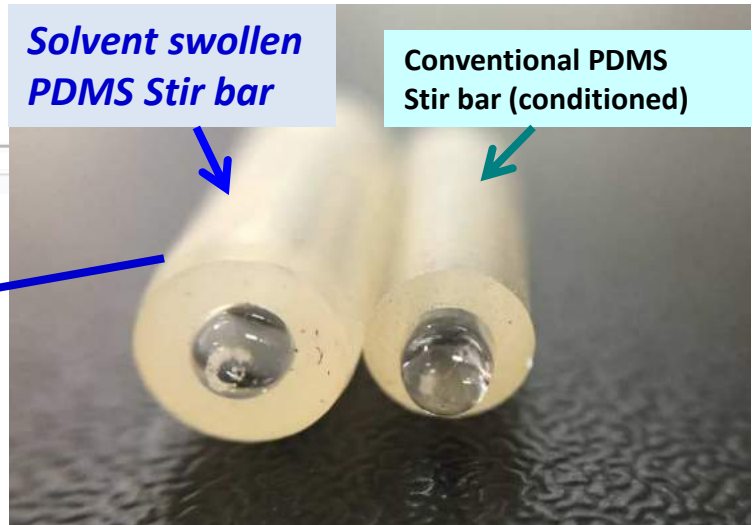
doi:10.1016/j.chroma.2016.05.085

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**SBSE using a solvent swollen PDMS stir bar**



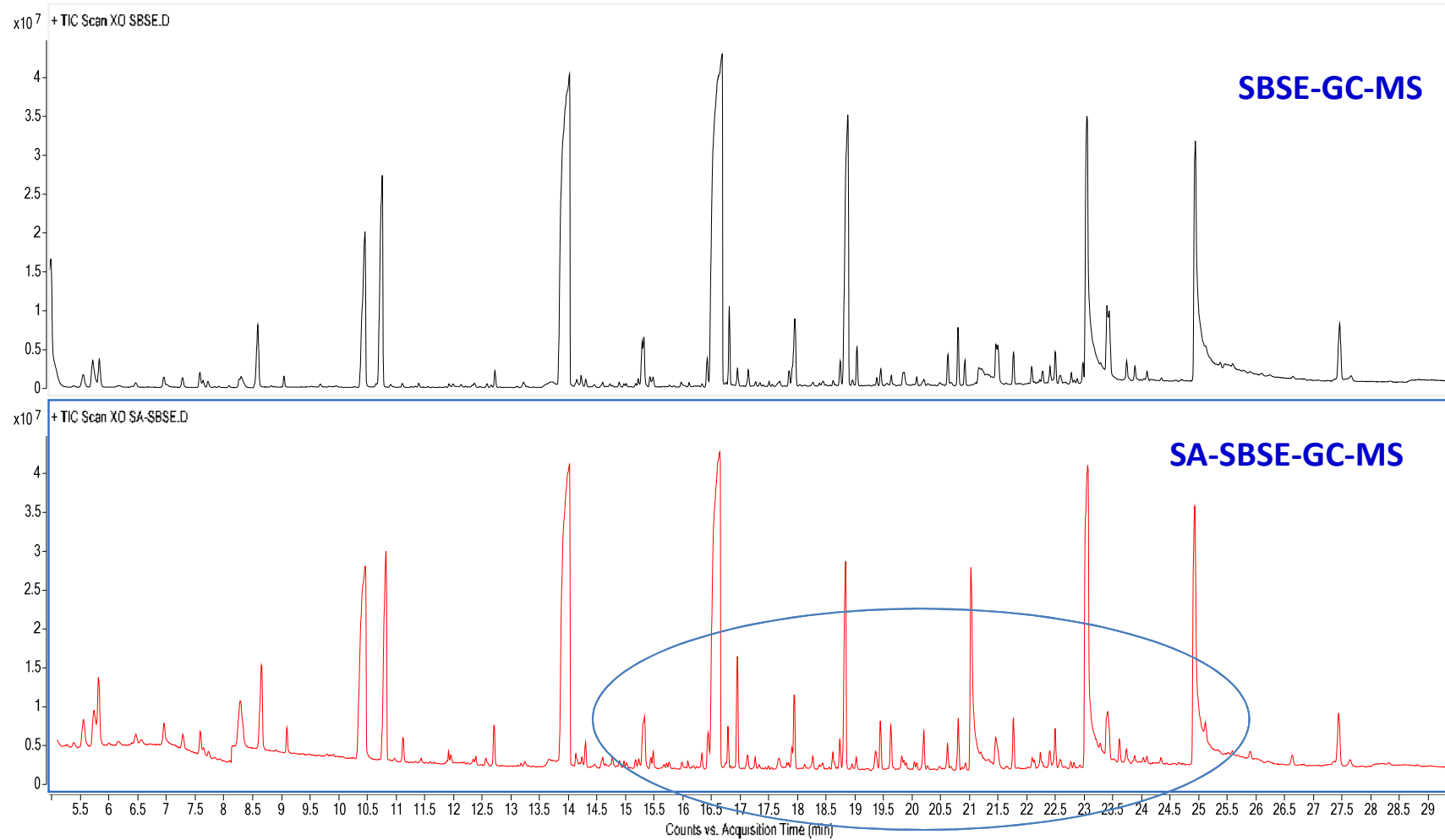
**SA-SBSE**



△ Solvent

● Solutes ( $\log K_{ow} < 2.0$ )

# SA-SBSE versus classical SBSE – Cognac XO



# Conclusions

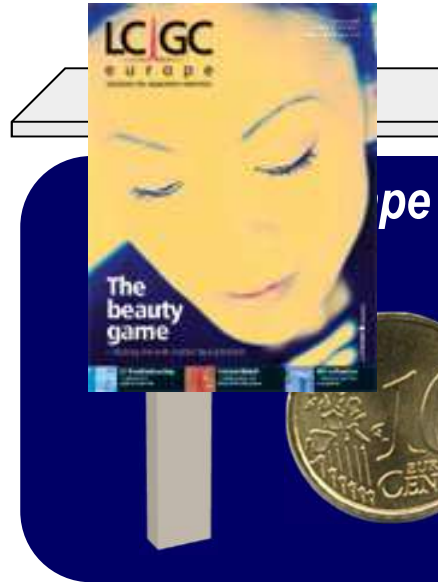
- **SBSE is a mature sample preparation technique: automated – miniaturized – solventless**
- **Very high enrichment factor**
- **Reduced risk of contamination**
- **On-site sampling/extraction**
- **Wide application area: QC – contaminants – metabolomics**

# Other sorptive extraction methods

- Thin film SPME
- PDMS membrane (patch) sampling

# Sorptive Extraction in Flavour and Fragrance Analysis

## Carlo Bicchi, SBSE Technical meeting 2017



**PDMS tape:**  
length: 1-2 cm  
width: 0.5-1 cm  
thickness: 0.25-1



**Thank you and  
enjoy the meeting**



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